organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Hui Liu,^a* Hong-Qing Wang^b and Zhao-Jie Liu^a

^aKey Laboratory of Pesticides and Chemical Biology, Ministry of Education, College of Chemistry, Central China Normal University, Wuhan 430079, People's Republic of China, and ^bCollege of Chemistry and Chemical Engineering, Nanhua University, Hengyang 421001, People's Republic of China

Correspondence e-mail: huiliu0920@hotmail.com

Key indicators

Single-crystal X-ray study T = 292 KMean $\sigma(\text{C}-\text{C}) = 0.008 \text{ Å}$ R factor = 0.063 wR factor = 0.185 Data-to-parameter ratio = 15.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Ethyl {4-[5-(3-methylphenyl)-3-methylsulfanyl-4oxo-1-phenylpyrazolo[3,4-*d*]pyrimidin-6-yloxy]phenoxy}acetate

The title compound, $C_{29}H_{26}N_4O_5S$, is based on an almost planar pyrazolo[3,4-*d*]pyrimidin-4-one core. Molecules are packed through weak intermolecular $C-H\cdots O$ and $C-H\cdots N$ interactions.

Received 13 October 2006 Accepted 1 November 2006

Comment

Pyrazolo[3,4-*d*]pyrimidin-4-one derivatives exhibit a wide spectrum of biological activity, including antibacterial, antiphlogistic and antitumour properties (Ali *et al.*, 2001; Armstrong *et al.*, 2000; EI-Bendary & Badria, 2000; Tetsuya *et al.*, 1996). The title compound, (I), may then be used as a precursor for obtaining new bioactive molecules. In this paper we present the X-ray crystallographic analysis of this compound.



As shown in Fig. 1, ring atoms in the pyrazolo[3,4-*d*]pyrimidin-4-one group form an essentially planar system. The C7—N2, C8—C9 and C12—N4 bond lengths of 1.313 (5), 1.374 (5) and 1.284 (5) Å, respectively, are longer than those of typical C—N (1.28 Å; Sasada, 1984; Chen & Jin, 2002) and C—C bonds (1.34 Å). In contrast, the single bond lengths C9–N1, C7–C8, C8–C11, C12–N3, and C9–N4, in the range 1.345 (5)–1.433 (5) Å, are significantly shorter than typical Csp^2 –N (1.426 Å) and C–C (1.53 Å) bond lengths, reflecting a degree of delocalization in the pyrazolo[3,4*d*]pyrimidin-4-one system.

Weak intermolecular C-H···O and C-H···N hydrogen bonds are present in the crystal packing (Table 1 and Fig. 2), as well as π - π and C-H··· π interactions. Thus, the dihedral angle between five-membered rings defined by atoms N1/N2/ C7/C8/C9 and the neighbouring symmetry-related pyrazole ring (symmetry code: 1 - y, x, $\frac{1}{4} + z$) is 4.77 (1)°, and the distance between the corresponding ring centroids is 3.775 (2) Å.

Experimental

To a solution of 4-(ethoxycarbonyl)-3-(methylsulfanyl)-1-phenyl-1*H*-pyrazolo-5-yl iminophosphorane (2 mmol), prepared according to the reported procedure (Molina *et al.*, 1998), in dry dichloromethane

All rights reserved

© 2006 International Union of Crystallography



Figure 1

The molecular structure of (I), showing the atom-labelling scheme, and with displacement ellipsoids at the 30% probability level.



Figure 2

The packing of (I), viewed along [100]. Weak hydrogen bonds are indicated by dashed lines.

(25 ml) was added 3-methylphenyl isocyanate (2 mmol) under N₂ at room temperature. After the reaction mixture had been stirred for 2– 5 h, the solvent was removed under reduced pressure, and then anhydrous acetonitrile (25 ml), 2-(4-hydroxyphenoxy)carboxylate (2.0 mmol) (Liu *et al.*, 2005) and anhydrous potassium carbonate (0.05 g) were added to the mixture. After refluxing for 5 h, the mixture was filtered and concentrated under reduced pressure. The resulting residue was recrystallized from ethanol to give (I) in 82% yield. Crystals suitable for X-ray diffraction were grown from CHCl₃ at 293 K.

Crystal data

C29H26N4O5S	
$M_r = 542.60$	
Tetragonal, P41	
a = 14.1908 (16) Å	
c = 13.760 (3) Å	
V = 2771.0 (8) Å ³	
Z = 4	

Data collection

Bruker SMART APEX CCD area-
detector diffractometer
φ and ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 2002)
$T_{\min} = 0.953, \ T_{\max} = 0.987$

Refinement

Refinement on F^2	
$R[F^2 > 2\sigma(F^2)] = 0.063$	
$wR(F^2) = 0.185$	
S = 0.95	
5408 reflections	
356 parameters	
H-atom parameters constrained	

 $D_x = 1.301 \text{ Mg m}^{-3}$ Mo $K\alpha$ radiation $\mu = 0.16 \text{ mm}^{-1}$ T = 292 (2) KPlate, colourless $0.30 \times 0.20 \times 0.04 \text{ mm}$

27946 measured reflections 5408 independent reflections 3258 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.131$ $\theta_{\text{max}} = 26.0^{\circ}$

$w = 1/[\sigma^2(F_o^2) + (0.12P)^2]$
where $P = (F_0^2 + 2F_c^2)/3$
$(\Delta/\sigma)_{\rm max} < 0.001$
$\Delta \rho_{\rm max} = 0.58 \ {\rm e} \ {\rm \AA}^{-3}$
$\Delta \rho_{\rm min} = -0.31 \text{ e } \text{\AA}^{-3}$
Absolute structure: Flack (1983),
2572 Friedel pairs
Flack parameter: -0.11 (13)

Table 1

Hydrogen-bond geometry (A, *

N TT - 4	D II	TT 4	D 4	
$D - H \cdots A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
$22 - H22 \cdots O1^{i}$	0.93	2.45	3.275 (6)	148
$25 - H25 \cdots O1^{ii}$	0.93	2.49	3.141 (5)	127
$-H6 \cdot \cdot \cdot N4$	0.93	2.32	2.969 (6)	126

Symmetry codes: (i) -y + 1, x - 1, $z + \frac{1}{4}$; (ii) y, -x + 1, $z - \frac{1}{4}$.

H atoms were included in calculated positions, with C–H distances constrained to 0.93 (aromatic CH), 0.96 (methyl CH₃) and 0.97 Å (methylene CH₂). Methyl groups were considered as rigid groups but were allowed to rotate freely about their C–C bonds. Isotropic displacement parameters for H atoms were fixed at $1.5U_{eq}$ (carrier atom) for methyl H atoms and $1.2U_{eq}$ (carrier atom) for other H atoms. Owing to unresolved disorder in the ethyl ester region, atoms C27, C28, C29, O4 and O5 were restrained so that their U^{ij} components approximated to isotropic behaviour.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 2001); software used to prepare material for publication: *SHELXTL*.

The authors are grateful to the Hubei Province Science Fund for Distinguished Young Scholars (2004ABB011) for support of this research. We thank Xiang-Gao Meng, Sheng-Zhen Xu and Xin-Yong Li for fruitful discussions.

References

Ali, A., Taylor, G. E. & Graham, D. W. (2001). Patent No. WO 0129045.

Armstrong, S. A., Berge, J. M., Brown, P., Elder, J. S., Forrest, A. K., Hamprecht, D. W. & Jarvest, R. L. (2000). Patent No. WO 0071524.

Bruker (2001). SMART (Version 5.628), SAINT (Version 6.45) and SHELXTL (Version 5.0). Bruker AXS Inc., Madison, Wisconsin, USA.

- Chen, W.-Q. & Jin, G.-Y. (2002). Phosphorus Sulfur Silicon Relat. Elem. 177, 1193–1200.
- EI-Bendary, E. R. & Badria, F. A. (2000). Arch. Pharm. 333, 99-103.
- Flack, H. D. (1983). Acta Cryst. A39, 876-881.
- Liu, H., Wang, H. Q. & Liu, Z. J. (2005). Chin. J. Appl. Chem. 23, 286–290.
- Molina, P., Alajarin, M., Vidal, A., Elguero, J. & Claramunt, R. M. (1998). *Tetrahedron*, **44**, 2249–2259.
- Sasada, Y. (1984). *Molecular and Crystal Structure in Chemistry Handbook*, 3rd ed. Tokyo: The Chemical Society of Japan, Maruzen.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Sheldrick, G. M. (2002). SADABS. Version 2.03. University of Göttingen, Germany.
- Tetsuya, A., Shogo, M., Fumio, I., Masuo, Y. & Masafumi, N. (1996). Patent No. EP 0733633.