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## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.008 \AA$
$R$ factor $=0.063$
$w R$ factor $=0.185$
Data-to-parameter ratio $=15.2$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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## Ethyl \{4-[5-(3-methylphenyl)-3-methylsulfanyl-4-oxo-1-phenylpyrazolo[3,4-d]pyrimidin-6-yloxy]phenoxy\}acetate

The title compound, $\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{~S}$, is based on an almost planar pyrazolo[3,4- $d$ ]pyrimidin-4-one core. Molecules are packed through weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-$ $\mathrm{H} \cdots \mathrm{N}$ interactions.

## Comment

Pyrazolo[3,4- $d$ ]pyrimidin-4-one derivatives exhibit a wide spectrum of biological activity, including antibacterial, antiphlogistic and antitumour properties (Ali et al., 2001; Armstrong et al., 2000; EI-Bendary \& Badria, 2000; Tetsuya et al., 1996). The title compound, (I), may then be used as a precursor for obtaining new bioactive molecules. In this paper we present the X-ray crystallographic analysis of this compound.


As shown in Fig. 1, ring atoms in the pyrazolo[3,4-d]-pyrimidin-4-one group form an essentially planar system. The $\mathrm{C} 7=\mathrm{N} 2, \mathrm{C} 8=\mathrm{C} 9$ and $\mathrm{C} 12=\mathrm{N} 4$ bond lengths of 1.313 (5), 1.374 (5) and 1.284 (5) Å, respectively, are longer than those of typical $\mathrm{C}=\mathrm{N}(1.28 \AA$; Sasada, 1984; Chen \& Jin, 2002) and $\mathrm{C}=\mathrm{C}$ bonds $(1.34 \AA)$. In contrast, the single bond lengths $\mathrm{C} 9-\mathrm{N} 1, \mathrm{C} 7-\mathrm{C} 8, \mathrm{C} 8-\mathrm{C} 11, \mathrm{C} 12-\mathrm{N} 3$, and $\mathrm{C} 9-\mathrm{N} 4$, in the range 1.345 (5)-1.433 (5) $\AA$, are significantly shorter than typical $\mathrm{Csp}^{2}-\mathrm{N}(1.426 \AA)$ and $\mathrm{C}-\mathrm{C}(1.53 \AA)$ bond lengths, reflecting a degree of delocalization in the pyrazolo[3,4$d$ ]pyrimidin-4-one system.

Weak intermolecular $\mathrm{C}-\mathrm{H} \cdots \mathrm{O}$ and $\mathrm{C}-\mathrm{H} \cdots \mathrm{N}$ hydrogen bonds are present in the crystal packing (Table 1 and Fig. 2), as well as $\pi-\pi$ and $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions. Thus, the dihedral angle between five-membered rings defined by atoms $\mathrm{N} 1 / \mathrm{N} 2 /$ $\mathrm{C} 7 / \mathrm{C} 8 / \mathrm{C} 9$ and the neighbouring symmetry-related pyrazole ring (symmetry code: $1-y, x, \frac{1}{4}+z$ ) is $4.77(1)^{\circ}$, and the distance between the corresponding ring centroids is 3.775 (2) Å.

## Experimental

To a solution of 4-(ethoxycarbonyl)-3-(methylsulfanyl)-1-phenyl-1 H -pyrazolo-5-yl iminophosphorane ( 2 mmol ), prepared according to the reported procedure (Molina et al., 1998), in dry dichloromethane

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Figure 1
The molecular structure of (I), showing the atom-labelling scheme, and with displacement ellipsoids at the $30 \%$ probability level.


Figure 2
The packing of (I), viewed along [100]. Weak hydrogen bonds are indicated by dashed lines.
( 25 ml ) was added 3-methylphenyl isocyanate ( 2 mmol ) under $\mathrm{N}_{2}$ at room temperature. After the reaction mixture had been stirred for $2-$ 5 h , the solvent was removed under reduced pressure, and then anhydrous acetonitrile ( 25 ml ), 2-(4-hydroxyphenoxy)carboxylate ( 2.0 mmol ) (Liu et al., 2005) and anhydrous potassium carbonate $(0.05 \mathrm{~g})$ were added to the mixture. After refluxing for 5 h , the mixture was filtered and concentrated under reduced pressure. The resulting residue was recrystallized from ethanol to give (I) in $82 \%$ yield. Crystals suitable for X-ray diffraction were grown from $\mathrm{CHCl}_{3}$ at 293 K .

## Crystal data

$\mathrm{C}_{29} \mathrm{H}_{26} \mathrm{~N}_{4} \mathrm{O}_{5} \mathrm{~S}$
$M_{r}=542.60$
Tetragonal, $P 4_{1}$
$a=14.1908$ (16) £
$c=13.760$ (3) A
$V=2771.0(8) \AA^{3}$
$Z=4$
$D_{x}=1.301 \mathrm{Mg} \mathrm{m}^{-3}$
Mo $K \alpha$ radiation
$\mu=0.16 \mathrm{~mm}^{-1}$
$T=292$ (2) K
Plate, colourless
$0.30 \times 0.20 \times 0.04 \mathrm{~mm}$

## Data collection

Bruker SMART APEX CCD areadetector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 2002)
$T_{\text {min }}=0.953, T_{\max }=0.987$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.063$
$w R\left(F^{2}\right)=0.185$
$S=0.95$
5408 reflections
356 parameters
H -atom parameters constrained

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.12 P)^{2}\right] \\
& \quad \text { where } P=\left(F_{\mathrm{o}}^{2}+2 F_{\mathrm{c}}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }<0.001 \\
& \Delta \rho_{\max }=0.58 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.31 \mathrm{e}^{-3} \AA^{-3} \\
& \text { Absolute structure: Flack (1983), } \\
& 2572 \text { Friedel pairs } \\
& \text { Flack parameter: }-0.11(13)
\end{aligned}
$$

Table 1
Hydrogen-bond geometry $\left(\AA^{\circ},{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{C} 22-\mathrm{H} 22 \cdots \mathrm{O}^{\mathrm{i}}$ | 0.93 | 2.45 | $3.275(6)$ | 148 |
| $\mathrm{C} 25-\mathrm{H} 25 \cdots 1^{\mathrm{ii}}$ | 0.93 | 2.49 | $3.141(5)$ | 127 |
| $\mathrm{C} 6-\mathrm{H} 6 \cdots \mathrm{~N} 4$ | 0.93 | 2.32 | $2.969(6)$ | 126 |

Symmetry codes: (i) $-y+1, x-1, z+\frac{1}{4}$; (ii) $y,-x+1, z-\frac{1}{4}$.

H atoms were included in calculated positions, with $\mathrm{C}-\mathrm{H}$ distances constrained to 0.93 (aromatic CH ), 0.96 (methyl $\mathrm{CH}_{3}$ ) and $0.97 \AA$ (methylene $\mathrm{CH}_{2}$ ). Methyl groups were considered as rigid groups but were allowed to rotate freely about their $\mathrm{C}-\mathrm{C}$ bonds. Isotropic displacement parameters for H atoms were fixed at $1.5 U_{\text {eq }}$ (carrier atom) for methyl H atoms and $1.2 U_{\text {eq }}$ (carrier atom) for other H atoms. Owing to unresolved disorder in the ethyl ester region, atoms $\mathrm{C} 27, \mathrm{C} 28, \mathrm{C} 29, \mathrm{O} 4$ and O 5 were restrained so that their $U^{i j}$ components approximated to isotropic behaviour.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL.

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## References

Ali, A., Taylor, G. E. \& Graham, D. W. (2001). Patent No. WO 0129045. Armstrong, S. A., Berge, J. M., Brown, P., Elder, J. S., Forrest, A. K., Hamprecht, D. W. \& Jarvest, R. L. (2000). Patent No. WO 0071524.
Bruker (2001). SMART (Version 5.628), SAINT (Version 6.45) and SHELXTL (Version 5.0). Bruker AXS Inc., Madison, Wisconsin, USA.

## organic papers

Chen, W.-Q. \& Jin, G.-Y. (2002). Phosphorus Sulfur Silicon Relat. Elem. 177, 1193-1200.
EI-Bendary, E. R. \& Badria, F. A. (2000). Arch. Pharm. 333, 99-103.
Flack, H. D. (1983). Acta Cryst. A39, 876-881.
Liu, H., Wang, H. Q. \& Liu, Z. J. (2005). Chin. J. Appl. Chem. 23, 286290.

Molina, P., Alajarin, M., Vidal, A., Elguero, J. \& Claramunt, R. M. (1998). Tetrahedron, 44, 2249-2259.

Sasada, Y. (1984). Molecular and Crystal Structure in Chemistry Handbook, 3rd ed. Tokyo: The Chemical Society of Japan, Maruzen.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Sheldrick, G. M. (2002). SADABS. Version 2.03. University of Göttingen, Germany.
Tetsuya, A., Shogo, M., Fumio, I., Masuo, Y. \& Masafumi, N. (1996). Patent No. EP 0733633.


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